The PFAS Universe: Uses, Classification & Degradation

Webinar for IC2 & MA TURI

Steve Korzeniowski & Bob Buck

30 January 2019





Meeting Agenda

- PFAS Universe (50-60min)
 - Introduction
 - Description, Two Categories, Five Classes
 - Current Uses
 - TURI Focus
 - Q&A
- PFAS Degradation Pathways(20-30 min)
 - for PFAS Currently in Use Q&A
- Group Discussion (10 min)
 - on Webinar Content and Follow Up Items

PFAS: Uses, Occurrence, Messages

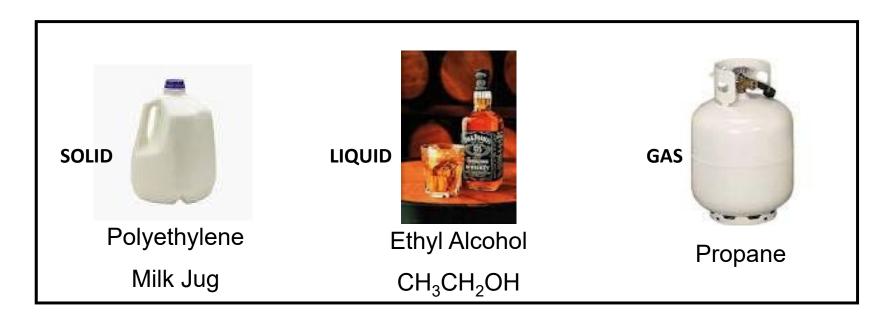
- In industrial and consumer products manufactured in U.S. and globally
- In manufacturing and processing
- Produced as intermediates and raw materials
- Occur as byproducts, as metabolites and via degradation

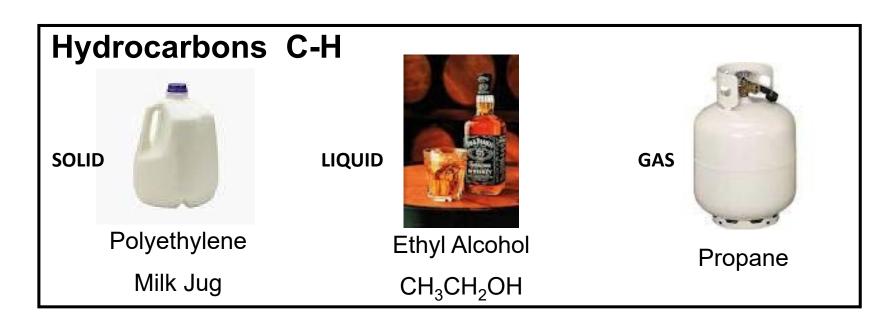
PFAS: BE CLEAR, SPECIFIC AND DESCRIPTIVE

PFAS Introduction Description, Two Categories, Five Classes

Aliphatic Hydrocarbons, C-H Substances

A Big Universe of Very Different Substances







SOLID



Fluorotelomer Alcohol **LIQUID** $C_nF_{2n+1}CH_2CH_2OH$

GAS



HFC-134a Refrigerant

Polytetrafluoroethylene **PTFE Plastic**

Fluorocarbons – Also, a Big Universe of Very Different Substances

WHY are we talking about PFAS?

- PFOS and PFOA found globally in the environment.
- A need to identify their origin where did they come from?
 - and substances that break down in the environment to form them - precursors.
- More broadly, on homologues, longer and shorter, and their origins.
 - and more recently, additional highly fluorinated substances found in the environment.
- The PFAS terminology was created to help define the group of relevant substances and establish a common lexicon.
 - it is very important to use common terms and to use terms that are <u>clear, specific</u> <u>and descriptive</u>.

PFAS

Per- and Poly- Fluoroalkyl Substances

The substances included in the term "PFAS" are a wide universe of substances with <u>very different</u> physical, chemical, environmental and biological properties.

- "PFAS" is a <u>very</u> broad term and should be used only when talking about <u>all</u> substances in the group.
 - It is very important to use common terms and to use terms that are <u>clear</u>, <u>specific and descriptive</u>.

PFAS Definition*

aliphatic substance that contain the perfluoroalkyl moiety

 C_nF_{2n+1}

Perfluoroalkyl and Polyfluoroalkyl Substances in the Environment: Terminology, Classification, and Origins

"highly fluorinated aliphatic substances that contain 1 or more C atoms on which all the H substituents have been replaced by F atoms, in such a manner that they contain the perfluoroalkyl moiety C_nF_{2n+1}–. These compounds are hereafter referred to as "perfluoroalkyl and polyfluoroalkyl substances" and denoted by the acronym PFASs..."

*IEAM **2011**, 7(4):513-541.

Open access: http://dx.doi.org/10.1002/ieam.258

PFAS – What's included? Not?

PFAS

Non-Polymer:

- Perfluoroalkyl
- Polyfluoroalkyl

Polymer:

- Fluoropolymer
- Perfluoropolyether Polymers
- Side-Chain Fluorinated Polymers

Fluorinated Substances that <u>are not</u> PFAS

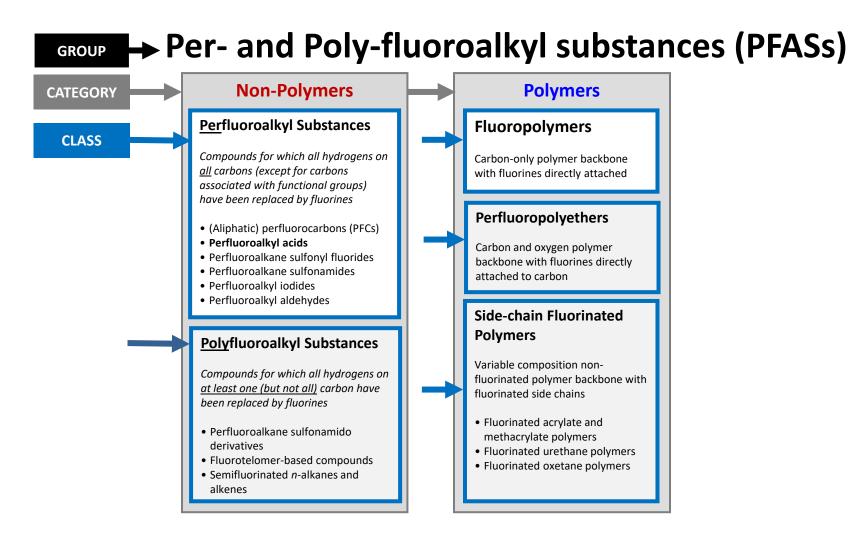
- Fluorochemicals
 - Refrigerants
- Fluorinated aromatics
 - hexafluorobenzene



Perfluoroalkyl and Polyfluoroalkyl Substances in the Environment: Terminology, Classification, and Origins

Open access:

http://dx.doi.org/10.1002/ieam.258



BE CLEAR, SPECIFIC AND DESCRIPTIVE

30 January 2019 11

PFAS - Uses

RM



TD

Raw Material

RM

- Telomer lodides
- Telomer Alcohols
- Perfluoroalkyl sulfonyl fluorides
- Tetrafluoroethylene

Commercial Product

C

- Polytetrafluoroethylene (PTFE)
- Side-chain fluorinated polymers
- Flurorosurfactants

Terminal
Degradation
Product

TD

PFCAs – e.g., PFOA PFSAs – e.g., PFOS

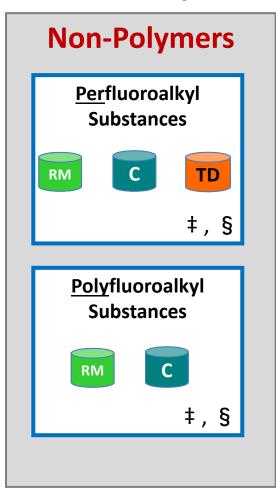
Per- and Poly-fluoroalkyl substances (PFASs)

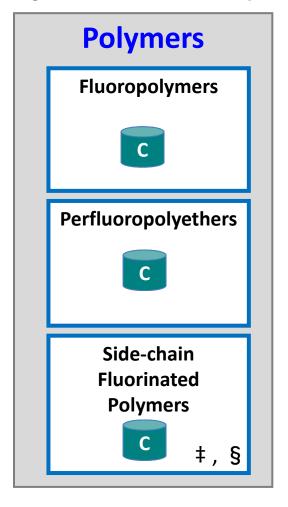
CATEGORY

GROUP

CLASS

- Raw Materials
- C Commercial Products
- Terminal
 Degradation
 Products





Main Manufacturing Technologies:

- ‡ Electrochemical Fluorination (ECF)
- § Fluorotelomer (FT)

GROUP

Per- and Poly-fluoroalkyl substances (PFASs)

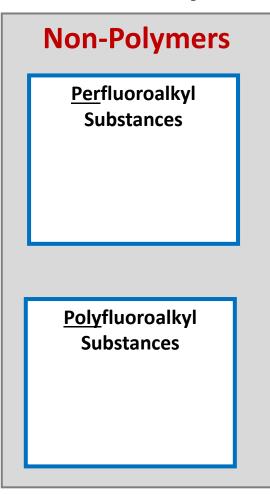
CATEGORY

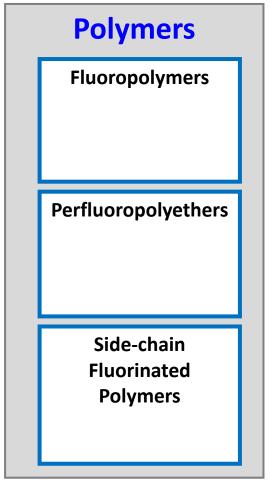
CLASS

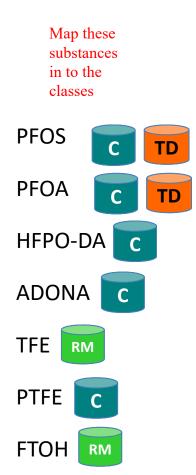
RM

C

TD







- FT Acrylate Polymer
- ECF Urethane Polymer

GROUP

Per- and Poly-fluoroalkyl substances (PFASs)

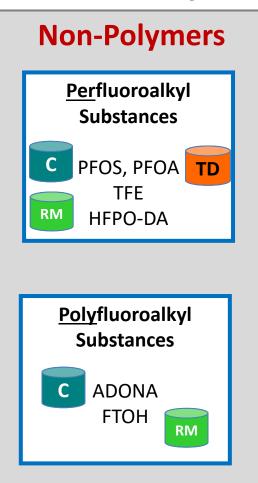
CATEGORY

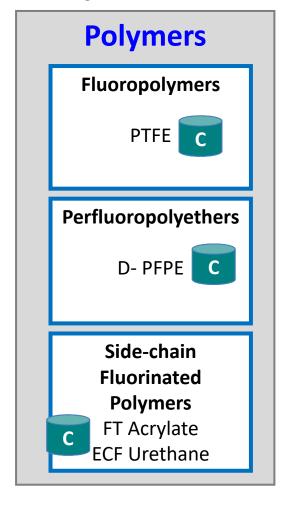
CLASS

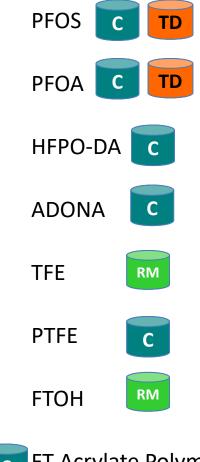
RM

C

TD







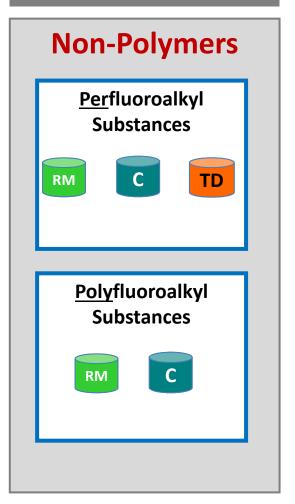
- ECF Urethane Polymer
- C D-PFPE

Non-Polymers

are a

Focus PFAS Category

FOCUS PFAS CATEGORY



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Specific Focus on

Perfluoroalkyl Acids -

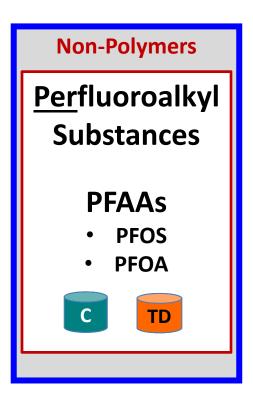
PFAAs

such as

PFOA and PFOS

And substances that may degrade in the environment to form them

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PFSA – Perfluoroalkane Sulfonic Acid (-SO₃H) e.g., **PFOS**, PFHxS, PFBS

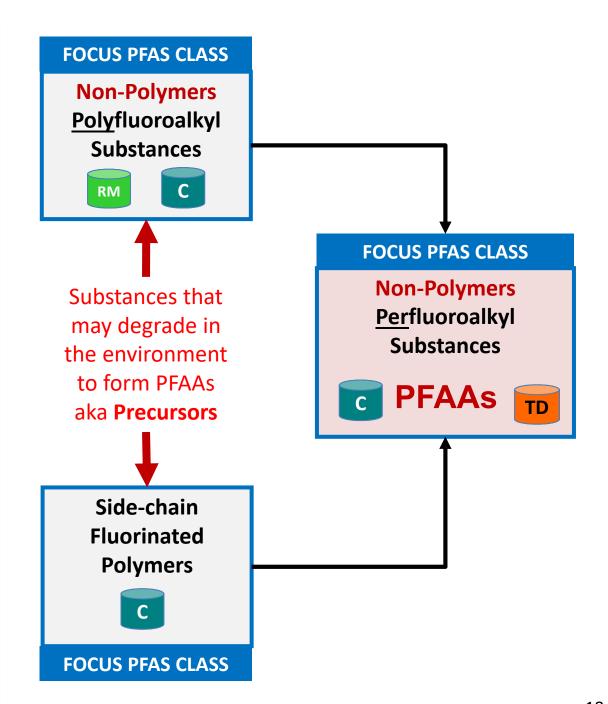
PFCA – Perfluorocarboxylic Acid (-CO₂H)

e.g., **PFOA**, PFHxA, PFBA

Focus on PFAAs

and

Substances that may degrade in the environment to form them



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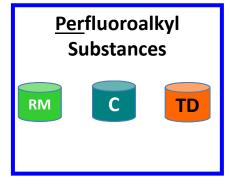
Non-Polymer Per-and Polyfluoroalkyl Substances

and

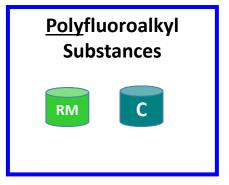
Side-Chain Fluorinated Polymers

are the Focus PFAS Classes

FOCUS PFAS CLASSES

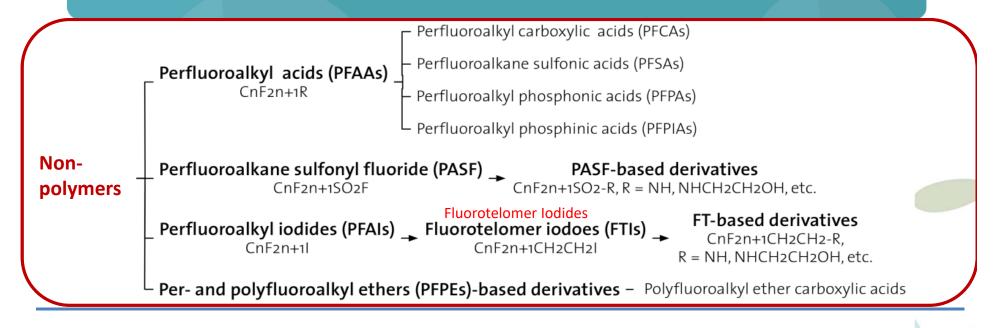


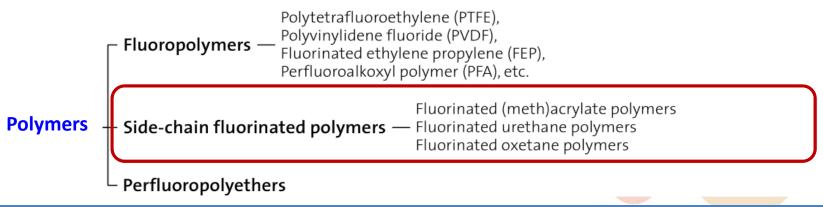
Side-chain Fluorinated Polymers



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OECD PFAS Map



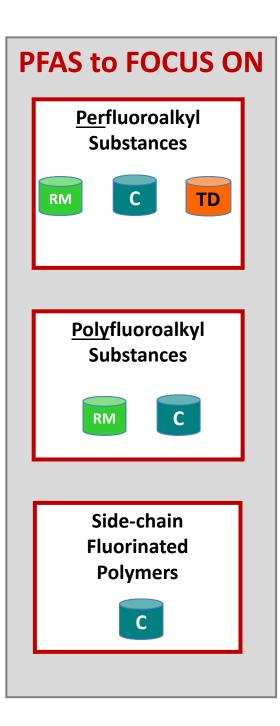


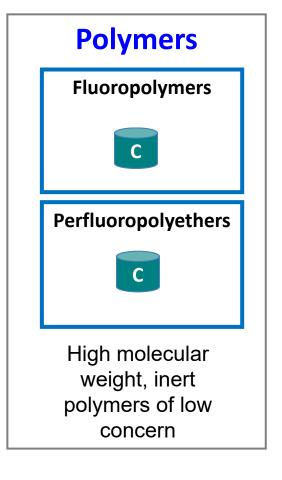
A New View

PFAS to FOCUS ON

including impurities and use of non-polymer PFAS such as PFECAs as PPAs in manufacturing Polymers

High molecular weight polymers themselves are not the FOCUS

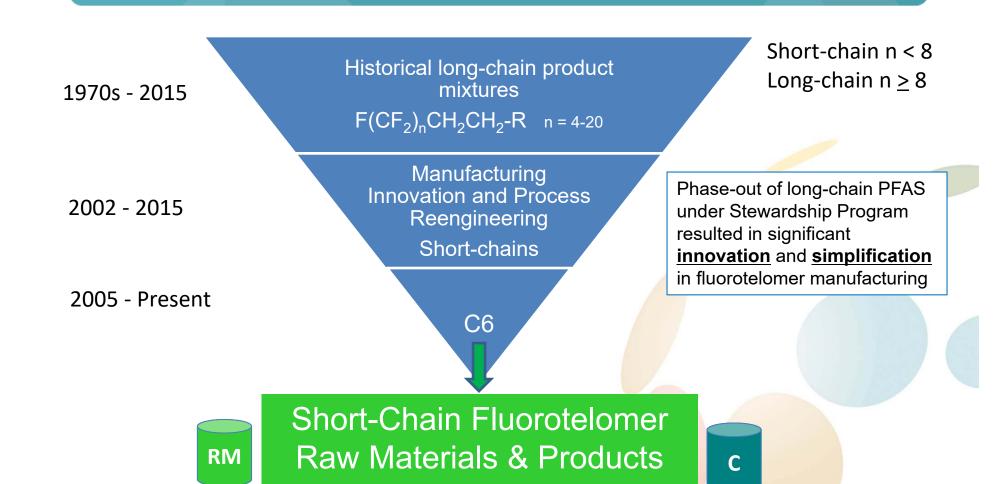




Specific topics

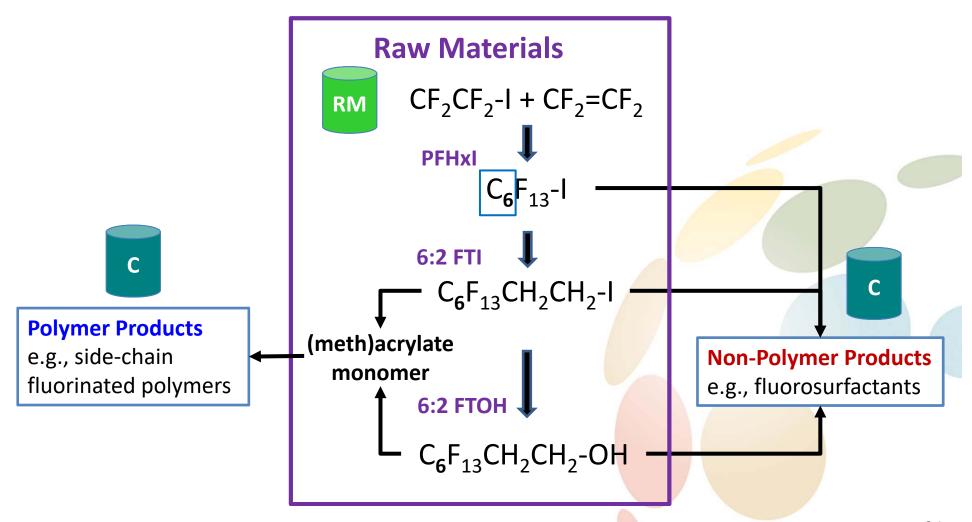
- Fluorotelomers
- ECF
- PPAs (polymer polymerization aids)
- Fluoropolymers
- PFPE Polymers

Fluorotelomer Manufacturing: Shift to Short-Chains

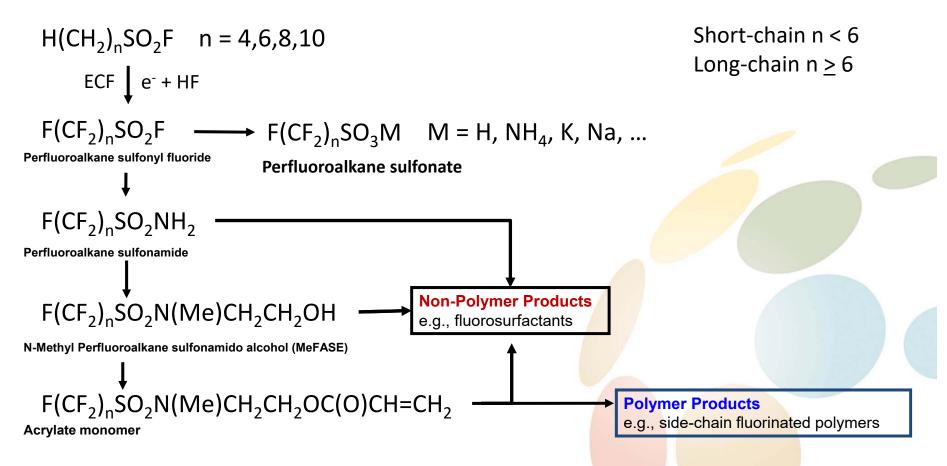


n = 4 or 6

Short-chain Fluorotelomers - C6 • • • • •



Electrochemical Fluorination (ECF)



ECF produced materials have up to 30% branched perfluoroalkyl chains and contain odd and even carbon chain lengths.

Polymer Polymerization Aids - PPA's

Per- and Polyfluoroalkyl ether carboxylates

PFAS CATEGORY

Non-Polymers

<u>Per</u>fluoroalkyl ether carboxylates



HFPO-DA;

JA;

HFPO-DA, CAS# 62037-80-3 F₃C-CF₂CF₂-O-CF(CF₃)-CO₂

① CAS# 908020-52-0 F₃C-CF₂-O-CF₂CG₂-

<u>Poly</u>fluoroalkyl ether carboxylates



ADONA;

ADONA, CAS# 958445-44 F₃C-O-CF₂CF₂CF₂-O-CFHCF₂-CO₂-

② CAS# 329238-24-6 CIC_3F_6 -O- $[CF_2CF(CF_3)O-]_n$ - $[CF(CF_3)-O-]_m$ - CF_2 - CO_2 -

Fluoropolymers

Integrated Environmental Assessment and Management — Volume 14, Number 3—pp. 316-334 Received: 26 September 2017 Returned for Revision: 16 January 2018

Accepted: 30 January 2018

http://dx.doi.org/10.1002/ieam.4035

Critical Review

316

A Critical Review of the Application of Polymer of Low Concern and Regulatory Criteria to Fluoropolymers

PTFE, FEP, ETFE, PFA, etc.

- Fluoropolymers have thermal, chemical, photochemical, hydrolytic, oxidative, and biological stability.
- Fluoropolymers are practically insoluble in water and not subject to long-range transport.
- With a molecular weight well over 100 000 Da, fluoropolymers cannot cross the cell membrane. Fluoropolymers are not bioavailable or bioaccumulative
- Clinical studies of patients receiving permanently implanted PTFE cardiovascular medical devices demonstrate no chronic toxicity or carcinogenicity and no reproductive, developmental, or endocrine toxicity.
- Degrade to form HF and CO₂ under municipal waste incineration conditions.

PFPE Polymers - Basics

- PFPE's are unique polymers which have excellent thermal stability, chemical inertness, as well as being liquids over a wide range of molecular weights
 - There are also products based on PFPE oligomers
- They are characterized as containing only Carbon, Fluorine and Oxygen, with differing amounts of Carbon and Oxygen
- Commercial products containing PFPEs are oils and greases used as lubricants where high thermal and chemical stability is critical for service and service lifetime.
 - Applications include:
 - Bearings
 - including Sealed for Life Applications
 - Vacuum Pump Fluids

Commercial Perfluoropolyethers - PFPEs

PFPE- K

$$F_3$$
C CF_2 CF_2 CF_3 CF_3

$$-Y \qquad F_3C \qquad CF_2 \qquad F \qquad CF_2 \qquad CF_3 \qquad CF_3$$

Perfluoroalkylpolyethers.

Synthetics, Mineral Oils, and Bio-Based Lubricants: Chemistry and Technology Rudnik, L. R., Ed. CRC Press, **2005**. pp 158-174



PFAS: Current Uses



Societal Value and Benefits of Fluorotechnology

- The use and variation of the carbon-fluorine bond gives FluoroTechnology its distinct properties of strength, durability, heatresistance and stability.
 - These properties are critical to the reliable and safe function of myriad products that industry and consumer rely on every day

PFAS Uses & Occurrence*

- In industrial and consumer products manufactured in U.S. and globally
- In manufacturing and processing
- Produced as intermediates and raw materials
- Occur as byproducts, as metabolites and via degradation

^{*}Sources: ITRC Fact Sheets; OECD Synthesis Paper; FluoroCouncil Infographic and Website

FluoroTechnology Benefits and Supported Industries

Durability

- Building and Construction
- Industrial and Institutional Floor Care
- Paints and Coatings
- Cookware
- Outdoor Apparel/Equipment
- Leather (Footwear and Furniture)
- Professional Textiles

Emissions Reduction/(Alternative) Energy

- Automotive, Trucks, Buses
- Oilfield and Mining
- Alternative Energy









FluoroTechnology Benefits and Supported Industries

Safety: Workers, Consumers, Environment

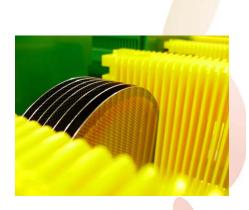
- Chemical and Pharmaceutical Manufacturing
- Food Processing
- Personal Protection and Worker Safety
- Safety Equipment
- Healthcare/Hospitals
- Metal Plating
- Aerospace
- Wire & Cable in Buildings
- Paper and Packaging

Electronic Equipment

Semiconductors and Electronics









FLUOROTECHNOLOGY MAKES IMPORTANT PRODUCTS FOR VITAL INDUSTRIES POSSIBLE

FluoroCouncil member companies voluntarily committed to a global phase-out of long-chain fluorochemistries by the end of 2015, resulting in the transition to alternatives, such as short-chain fluorochemistries that offer the same high-performance benefits, but with improved environmental and health profiles.

ELECTRONICS



Improves insulation, weather-ability, transparency and water-resistance. Provides smooth and smudge-resistant touch screens.



AEROSPACE/ DEFENSE

Enables chemical-resistant tubes, hoses and fluid seals; high and low temperature brake and hydraulic fluids used in aircraft control systems and brakes; and ultra-high frequency wire and cable insulation necessary for navigation, fly-by-wire control and aircraft communications.



BUILDING/ CONSTRUCTION

Enhances durability, UV resistance and anti-corrosive properties to lengthen the lifetime of infrastructure, facades and surfaces.

FIRST RESPONDERS

flammable liquid

fires.



AUTOMOTIVE



Provides every automotive system with durability, heat and chemical resistance and vapor barriers. Increases reliability of engine compartment wirings and gauges and improves auto safety by reducing engine compartment fires. Protects carpets and seats against stains, soil, oil and water.

ALTERNATIVE ENERGY

Enables lithium batteries, fuel cells and solar panels, which contribute to reduced emissions and energy costs.

OIL AND GAS

Provides reliable equipment to help improve the safety and affordability of oil-field and pipeline operations. Improves the reliability and safety of fuel system seals and hoses, O-rings and downhole and field equipment gaskets.

MILITARY



Enables apparel and equipment to provide high-barrier skin protection in extreme environments and against chemical warfare agents.

CHEMICAL/ PHARMACEUTICAL MANUFACTURING

Provides sterile. corrosion-resistant coatings, linings and equipment.

HEALTHCARE



Serves as high dielectric insulators in medical equipment that relies on high frequency signals, like defibrillators, pacemakers and CRT, PET and MRI imaging devices. Used to treat medical garments, drapes and divider curtains to protect against the transmission of diseases and

infections.

SEMICONDUCTORS

manufacturing environments necessary for micro-electronics. Used for plasma machinery, etching materials, cleaning fluids and wetting surfactants for chemical etchants.

OUTDOOR APPAREL/ **EQUIPMENT**



Creates breathable membranes and long-lasting finishes that provide water repellency, oil repellency, stain resistance and soil release with abrasion-resistant finishes for apparel and equipment.

FLUORINE 1 CARBON

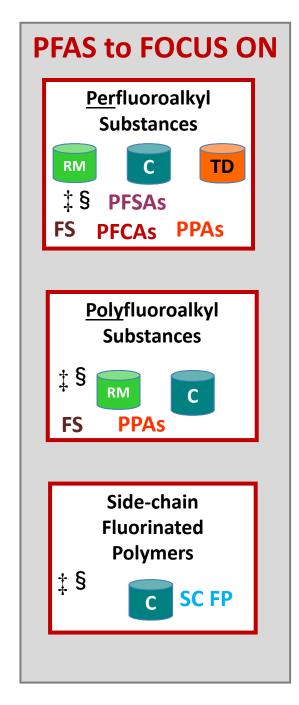
FluoroTechnology is the use of fluorine chemistry to create any fluorinated product. When fluorine and carbon atoms join together, they create a powerful chemical bond. The use and manipulation of this bond gives Fluoro Technology its distinct properties of strength, durability. heat-resistance and stability. These properties are critical to the reliable and safe function of myriad products that industry and consumer rely on every day.

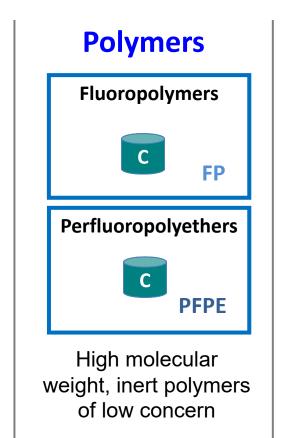


www.FluoroCouncil.org

Fluorotechnologies: Legend

- FP: Fluoropolymers
- PFPE: Perfluoropolyethers
- SC FP: Side-chain fluoropolymers
- FS: Fluorosurfactants
- PFCAs: Perfluoro carboxylic acids
- PFECAs: Perfluoroalkylether carboxylic acids (e.g., HFPO-DA)
 - Polyfluorolkyl ether carboxylic acids (e.g., ADONA)
- PFSAs: Perfluoroalkane sulfonic acids
- Terminal Degradation Products TD
- Raw Materials
- Commerical Products





- ‡ Electrochemical Fluorination (ECF)
- § Fluorotelomer (FT)

Commercial Products

- FP: Fluoropolymers
- PFPE: Perfluoropolyethers
- SC FP: Side-chain fluoropolymers
- FS: Fluorosurfactants
- PFCAs: Perfluoro carboxylic acids
- PPAs: PFECAs- Perfluoroalkylether carboxylic acids (e.g., HFPO-DA); Polyfluogolkyl ether carboxylic acids (e.g., ADONA)
- PFSAs: Perfluoroalkane sulfonic acids

Fluorotechnology Uses (1)

Electronics

- insulation, weatherability, transparency, waterresistance; smudge-resistant touch screens
- FP; PFPE

Aerospace/Defense/Aviation

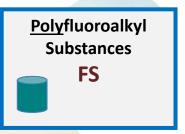
- chemical-resistant tubes & hoses, fluid seals; UHF wire & cable insulation; extreme temperature & conditions lubrication; brake & hydraulic fluids
- FP; PFPE; FS

Building/Construction

- exterior durability, UV resistance, anti-corrosive properties; wire & cable; paints & coatings
- FP; PFPE; FS



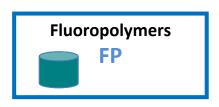




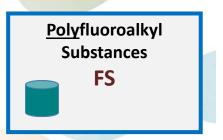
Fluorotechnology Uses (2)

First Responders

- protective turnout gear; AFFF for high-hazard
 Class B fires
- FP; FS; SC FP
- Alternative Energy
- lithium batteries, fuel cells, solar panels;
 membranes & protective coatings
- FP; PFPE; FS
- Automotive
- heat & chemical resistant tubes, hoses, seals, wiring; carpet & seat protection (see next chart)

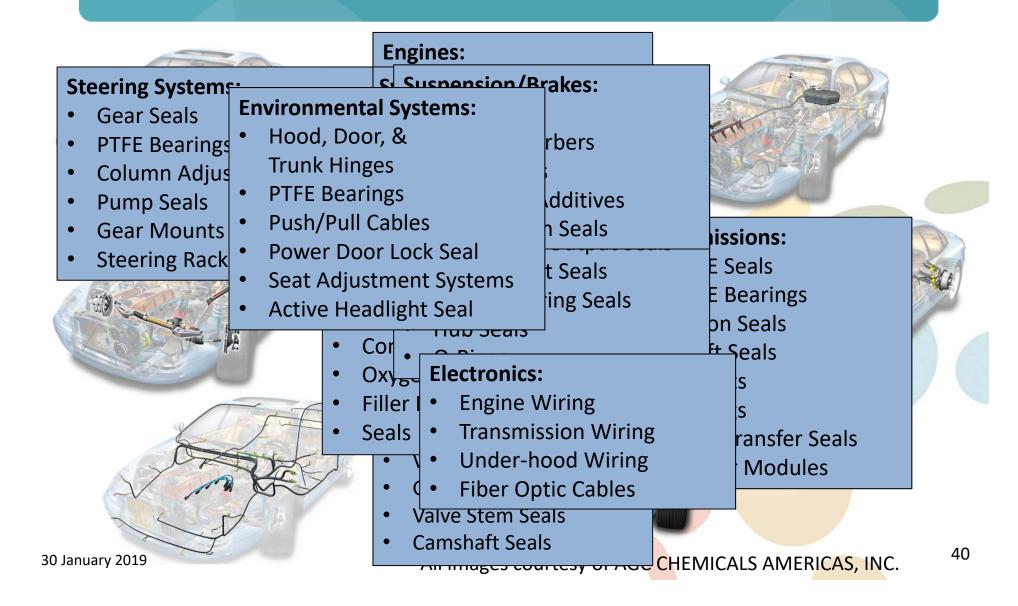








Automotive Uses of Fluoropolymers



Fluorotechnology Uses (3)

Military

- protective gear/armor, helmets and equipment for extreme environments and against chemical warfare agents; AFFF for Class B fires
- FP; FS; SC FP

Semiconductors

- ultra pure environments; etching materials; cleaning fluids; wetting surfactants; lubrication
- FP; PFPE; FS

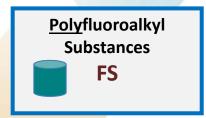
Oil & Gas

- equipment for oil-field, downhole and pipelines; fuel system seals, hoses, O-rings, gaskets; oil well and NG stimulation; lubrication
- FP; PFPE; FS









Fluorotechnology Uses (4)

Healthcare & Medical

 high dielectric insulators in defibrillators, pacemakers and imaging devices; surface protection treatments for medical garments, drapes



- FP; SC FP

Chemical/Pharma/Food Processing/General Manufacturing

 sterile, corrosion-resistant coatings, linings and equipment; lubrication in extreme environments



– FP; PFPE

Outdoor Apparel & Equipment

• breathable membranes and long-lasting finishes for high-performance uses: water/oil/stain/soil resistance

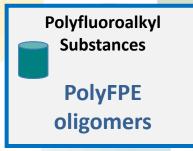


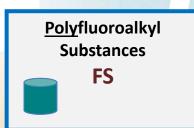
– FP: SC FP

Fluorotechnology Uses (5)

- Paper & Packaging
 - surface coatings for oil and grease repellency / protection in food-contact uses and non-food applications: boxes, masking papers, carbonless forms
 - SC FP; PolyFPE oligomers
- I&I and HH Cleaning Products
 - provides wetting, leveling in floor polishes and other cleaning products
 - FS; SC FP
- Biocides & Pesticides: used as active ingredient in pesticides, formulations
 - FS (ECF only)







Fluorotechnology Uses (6)

Polymer Polymerization

- used as a polymerization surfactant to make various high molecular weight fluoropolymers
- PFCAs; PFECAs; FS

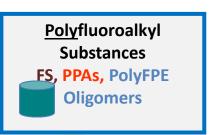
Metal Plating & Etching

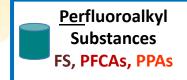
 corrosion prevention; wetting agent; fume suppression in electroplating for chrome, copper, nickel and tin

- FS

Carpet, Textile, Leather, Apparel

- coatings applied to provide water, oil, soil and stain repellency and in some cases impart soil release properties
- SC FP; PolyFPE oligomers



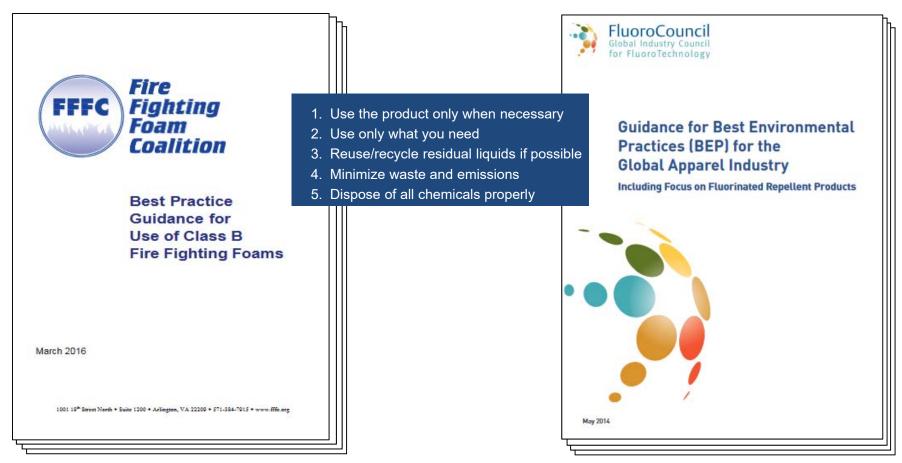




Fluorotechnolgy Uses & PFAS Classification

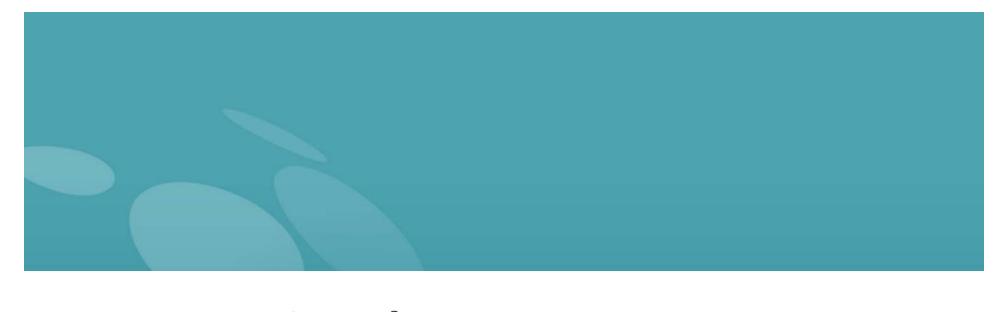
Use Areas	/	Classification				NP = N	on-polyı	mer		
Electronics		FP: Fluoropolymers			PFPE:	PFPE: Perfluoropolyethers				
Aerospace/Aviation		FS: Fluorosurfactants ((NP)	FP	PFPE			
Building/Construction		FP	PFPE	FS						
First Responders		FP	FS	SC FP: Side-chain fluoropolymers						
Automotive		FP	SC FP							
Alternative Energy		FP	PFPE	FS						
Military		FP	SC FP	FS			2			
Semiconductors		FP	PFPE	FS			1			
Oil & Gas		FP	PFPE	FS						
Healthcare/Medical		FP	SC FP							
Chemical/Pharma/Food Processing		FP	PFPE							
Outdoor Apparel & Equipment		FP	SC FP				1		11/19/20	
Paper & Packaging		SC FP PolyPFPE Oligo			omers		/			
I&I and HH Cleaning		FS	SC FP							
Biocides & Pesticides		FS (ECF)				A				
Polymer Polymerization	PFECAs: Perfluoroalkylether carboxylic acids (e.g., HFPO-DA) (NP)									
	Polyfluorolkyl ether carboxylic acids (e.g., ADONA)									
		PFCAs: Perfluoro carboxylic acids (NP) FS								
Metal Plating & Etching		FS			_					
Carpet/Textile/Leather/Apparel		SC FP	PFPE							

Industry Best Practices - Reduces Environmental Release and Potential For Exposure



Links: https://fluorocouncil.com/PDFs/Guidance-for-Best-Environmental-Practices-BEP-for-the-Global-Apparel-Industry.pdf https://fluorocouncil.com/PDFs/Best-Practice-Guide-for-Use-of-Class-B-Fire-Fighting-Foams-PDF.pdf

30 January 2019 46



PFAS Universe: TURI Focus



TURI Request for Information

- TURI Current Focus:

- Non-Polymer: Fluoroethers and polymerization aids
 - » Perfluoroalkyl (e.g. HFPO-DA)
 - » Polyfluoroalkyl (e.g., Adona®)
- Non-Polymer, PFAAs:
 - » Perfluoroalkyl Phosphonic Acids (PFPAs)
 - » Perfluoroalkyl Phosphinic Acids (PFPiAs)

EPA's new chemicals program

Discuss EPA's PMN/TSCA program and approvals

PFPAs and PFPiAs - Primer

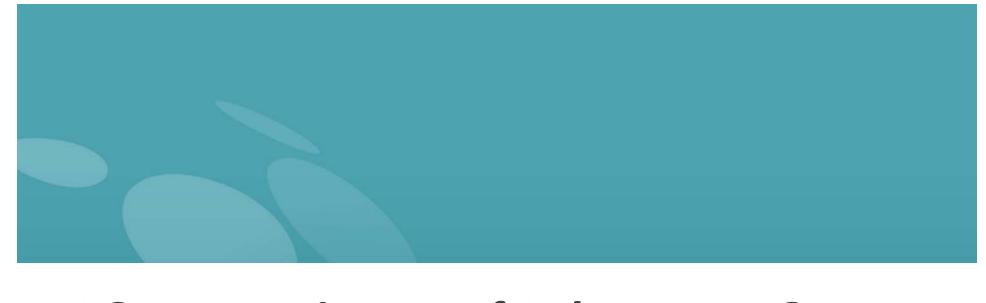
- Less well known, studied and used acids
- Manufactured by multiple companies historically
- Current manufacture and uses expected to be minor, if at all
- Reported to be used as defoamers in pesticide formulations and wetting agents in some consumer products*
- Structures:

Perfluoroalkyl Phosphonic Acids (PFPAs) $C_nF_{2n+1}PO(OH)_2$

Perfluoroalkyl Phosphinic Acids (PFPiAs) $(C_nF_{2n+1})(C_mF_{2m+1})PO(OH)$

EPA's PFAS PMN/TSCA Approval Process*

- Historically EPA took an integrated testing approach that involved a matrix of data needs
- Testing and data submission were required before PFAS alternatives (i.e. short chains)PMNs were approved
 - Work required on notified substance and/or degradant
- EPAs hazard and fate requirements included
 - Health and metabolism and Environmental fate
 - Half-life in animals and Human biomonitoring
- EPA issued Consent Orders (5e) that required testing on alternatives and are still effective and enforceable
 - Alternatives testing continues even today



Q&A on Universe of Substances & Uses



TURI Request for Information

- Global restriction limits, agreements, actions, residuals, exemptions,
 regulations and/or proposals review/highlight aka REACH and others
- TURI has been using May 2018 OECD diagram of PFAS compounds from: TOWARD A NEW COMPREHENSIVE GLOBAL DATABASE OF PER- AND POLYFLUOROALKYL SUBSTANCES (PFASs): SUMMARY REPORT ON UPDATING THE OECD 2007 LIST OF PER- AND POLYFLUOROALKYL SUBSTANCES (PFASs). Review current work plan with OECD to redefine/classify/broaden published PFAS list as noted
- Discuss request for annotated OECD spreadsheet of substances indicating which CAS #s/specific chemicals are in use in US and for which uses.
 There are approx. 100 CAS #'s that indicate reported in 2016 TSCA inventory. Note part of effort in b. Work in progress. Timing?

Degradation Pathways for PFAS Currently in Use



TURI Ask for Information

- Perfluoropolyether polymers, fluoropolymers, sidechain fluorinated polymers, fluorosurfactants: perand polyfluoroalkyl
- Experimental degradation data for chemicals in use and intermediates
 - Half-lives and rough percentages of breakdown products
 - Ultimate breakdown products
 - how many and which compounds are terminal end products?
 - Biotic pathways
 - Abiotic pathways (hydrolysis, oxidation or photolysis)

Biotic

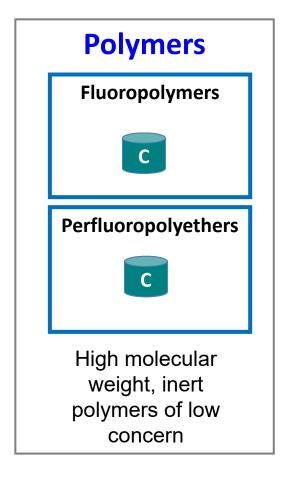
aerobic [O]

anaerobic

Abiotic
thermal (heat)
light
chemical
mechanical
hydrolysis

reasonably foreseeable use and environmental conditions

PFAS to FOCUS ON Perfluoroalkyl Substances RM ‡,§ **Poly**fluoroalkyl **Substances** RM Side-chain **Fluorinated Polymers**



Consider each PFAS class

Incineration (thermal)

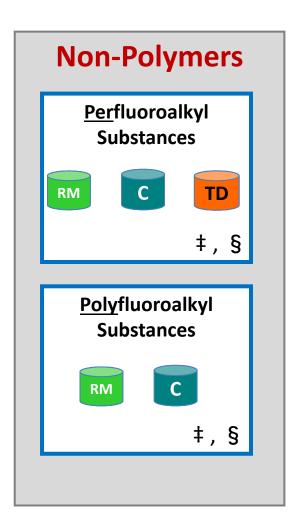
reasonably foreseeable use and environmental conditions

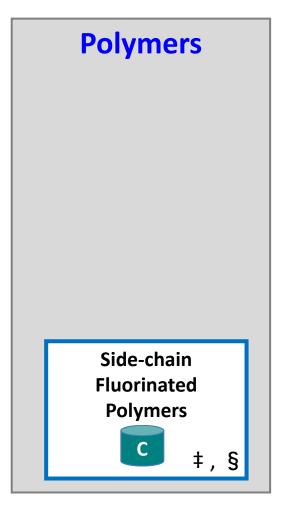
 Under typical municipal or medical waste incineration conditions, PFAS substances are destroyed forming HF and CO₂.

Fluorotelomerbased substances

ECF-based substances

reasonably foreseeable use and environmental conditions





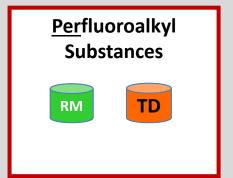
Manufacturing Technologies:

- ‡ Electrochemical Fluorination (ECF)
- § Fluorotelomer (FT)

Fluorotelomerbased substances

reasonably foreseeable use and environmental conditions

PFAS to FOCUS ON



Polyfluoroalkyl
Substances

Side-chain Fluorinated Polymers

Examples

Perfluorohexyl Iodide - PFHxI PFHxA TD



RM

- 6:2 Fluorotelomer Iodide
- 6:2 Fluorotelomer Alcohol (FTOH)
- 6:2 Fluorotelomer surfactants



6:2 Fluorotelomer-based polymers



Short-chain Fluorotelomer – C6 Biodegradation Pathway





Short-chain Fluorotelomer-based Products and Raw Materials



Transient Biodegradation Intermediates

e.g., 6:2 FTOH, 6:2 FTSA, 6:2 FTCA, 6:2 FTUCA, 5:3 Acid





Short-chain PFCA Terminal Degradation Products

PFCAs

PFBA

PFPeA

PFHxA

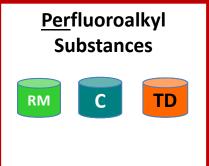
PFHpA

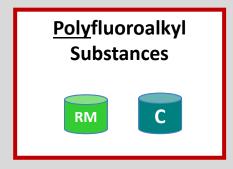
Degradation products formed and yields vary by substance, matrix (e.g., soil, sediment, air,) and mode (biotic vs. abiotic)

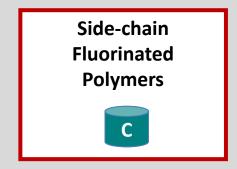
ECF-based substances

reasonably foreseeable use and environmental conditions

PFAS to FOCUS ON







Examples

Perfluorooctanesulfonyl fluoride (POSF)



PFOS PFBS





nEtFOSE

ECF Surfactants

C

Perfluorobutanesulfonamidobased acrylic polymers

C

Short-chain ECF Biodegradation Pathway





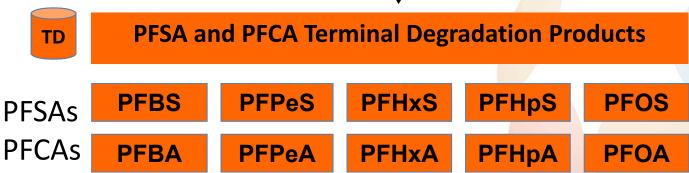
ECF-based Products and Raw Materials



Transient Biodegradation Intermediates

e.g., FASA, FASE, FASAA, etc.

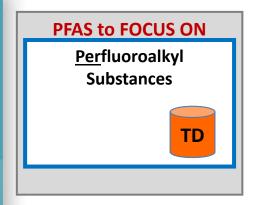




Degradation products formed and yields vary by substance, matrix (e.g., soil, sediment, air,) and mode (biotic vs. abiotic)

Perfluoroalkyl Acids - PFAAs

reasonable foreseeable use and environmental conditions



PFCA carboxylate

PFSA

PFPA

PFPiA phosphinate

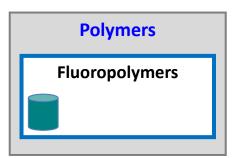
sulfonate

phosphonate

Perfluoroalkyl acids are not known to degrade in the environment or metabolize in living systems.

Fluoropolymers are stable and do not degrade

under reasonable foreseeable use and environmental conditions



- Fluoropolymers have thermal, chemical, photochemical, hydrolytic, oxidative, and biological stability.
- Fluoropolymers are practically insoluble in water and not subject to long-range transport.
- With a molecular weight well over 100 000 Da, fluoropolymers cannot cross the cell membrane. Fluoropolymers are not bioavailable or bioaccumulative
- Clinical studies of patients receiving permanently implanted PTFE cardiovascular medical devices demonstrate no chronic toxicity or carcinogenicity and no reproductive, developmental, or endocrine toxicity.
- Degrade to form HF and CO₂ under municipal waste incineration conditions.

A Critical Review of the Application of Polymer of Low Concern and Regulatory Criteria to Fluoropolymers.

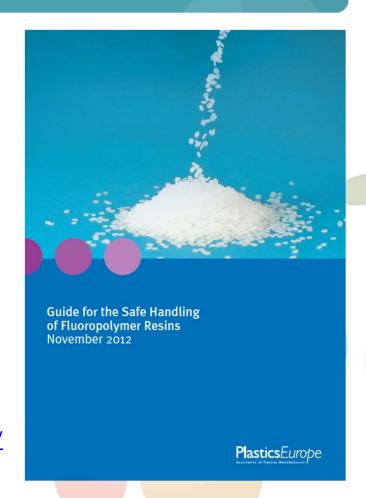
Integrated Environmental Assessment and Management

2018, 14(3):316-334. http://dx.doi.org/10.1002/ieam.4035

Fluoropolymers

- Proper use and handling
 - "Polymer Fume Fever"
 - Adequate ventilation must be used during industrial use and processing
 - Do not heat above decomposition temperature

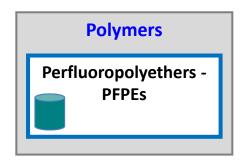
https://www.plasticseurope.org/en/resources/publications/ 108-guide-safe-handling-fluoropolymer-resins



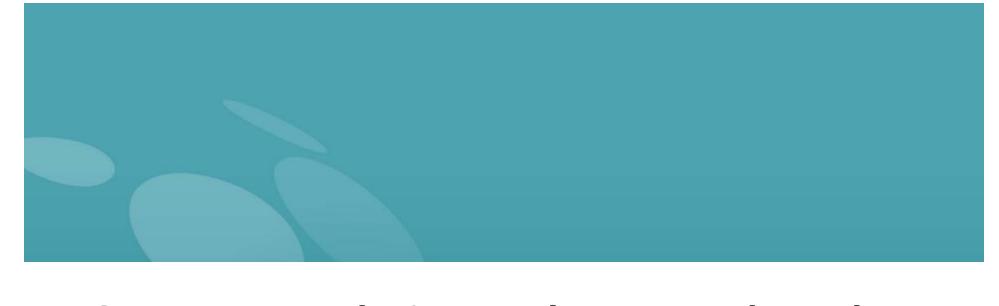
Perfluoropolyethers

PFPEs

reasonably foreseeable use and environmental conditions



- Perfluoropolyether polymers
 (PFPEs) have thermal, chemical, photochemical, hydrolytic, oxidative, and biological stability.
- PFPEs are practically insoluble in water and hydrocarbons and not subject to long-range transport.



Q&A on Degradation Pathways and Products



PFAS: Uses, Occurrence, Messages

- In industrial and consumer products manufactured in U.S. and globally
- In manufacturing and processing
- Produced as intermediates and raw materials
- Occur as byproducts, as metabolites and via degradation

PFAS: BE CLEAR, SPECIFIC AND DESCRIPTIVE

Group Discussion on Webinar Content and Follow Up Items

Thank you

See you at the ITRC annual meeting PFAS Workgroup in Boston March 26-28.